Semester	Course Code	<b>Course Category</b>	Hours/	Credits	Marks for Evaluation		
Semester	Course Coue	Course Category	Week		CIA	ESE	Total
III	24UCHVAC1	Value Added Course-I	30	-	-	100	100

# Course Title LABORATORY REAGENTS PREPARATION

SYLLABUS				
Unit	Contents			
Ι	Concentration of Solutions and their related terms: Standard solution. Concentration terms – Normality, Molarity, Molality, Formality, Mole fraction and Parts Per Million.	6		
II	<b>Equivalent Weight</b> Calculation of equivalent weight of HCl, H <sub>2</sub> SO <sub>4</sub> , H <sub>2</sub> C <sub>2</sub> O <sub>4</sub> , FAS, FeSO <sub>4</sub> , K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub> , H <sub>3</sub> PO <sub>3</sub> , H <sub>3</sub> PO <sub>4</sub> , CaCO <sub>3</sub> , NaOH, KOH, Ca(OH) <sub>2</sub> , Al(OH) <sub>3</sub> , KCl, AlCl <sub>3</sub> and KMnO <sub>4</sub> (Acid, Base and neutral solution).	6		
III	Preparation of Different Concentration Solutions - PracticalPreparation of N, N/2, N/5 and N/10 solution of HCl, H2SO4, H2C2O4, NaOH, KCl,KMnO4, AlCl3 and K2Cr2O7 in 100, 250, 500 and 1000 mL.	6		
IV	<b>Preparation of Parts Per Million Solutions - Practical</b> Preparation of 1, 10, 100, 250, 500 and 1000 ppm solution of CaCO <sub>3</sub> and NaCl for different volumes (Ex. 100, 250 and 500 mL).	6		
V	<b>Dilution Factor and Stock solution - Practical</b> Conversion of molar solution to normal solution and normal solution to molar solution. <b>Stock solution</b> – Preparation of 1N of HCl and 1N of H <sub>2</sub> C <sub>2</sub> O <sub>4</sub> for 50, 100, 250 and 500 mL from the stock.	6		
Text B	ook(s): Furniss, A.J. Hannaford, P.W.G. Smith, A.R. Tatchell, Vogel's Text Book of Practica	1.0		

1. B.S. Furniss, A.J. Hannaford, P.W.G. Smith, A.R. Tatchell, Vogel's Text Book of Practical Organic Chemistry, 5<sup>th</sup> Edition, Pearson Publications, India, 2019.

2. R. Gopalan and K.S. Viswanathan, Analytical Methods, Universities Press, India, 2018.

## **Reference Book(s):**

1. R.C. Mukerjee, Modern Approach to Chemical Calculations, Paperback Edition, Bharathi Bhawan Publishers, India, 2021.

### Web Resource(s):

1. https://en.wikipedia.org/wiki/Qualitative inorganic analysis

2. https://chem.libretexts.org/Bookshelves/Introductory\_Chemistry/Introductory\_Chemistry/13%3A\_Solutions/13.07%3A

## **Course Coordinator(s):**

Dr. M. Yaseen Mowlana, Dr. S. Farook Basha, Dr. G. Hema Sindhuja

Semester	Course Code	Course Cotogony	Hours/	Credits	Marks for Evaluation		
Semester	Course Coue	Course Category	Week	Creans	CIA	ESE	Total
V	24UCHVAC2	Value Added Course-II	30	-	-	100	100
Course Title THIN LAYER CHROMATOGRAPHY AND PHYSICAL CONSTANTS							TANTS

UnitContentsHourIThin Layer Chromatography 1.1 Thin Layer Chromatography-Principle, techniques, preparation of chromatoplates, application of sample on the chromatoplates. 1.2 Choice of adsorbents, selection of solvents, locating reagents, developing6	SYLLABUS				
1.1 Thin Layer Chromatography-Principle, techniques, preparation of chromatoplates, application of sample on the chromatoplates.6	Unit	Contents	Hours		
chamber, development types, location of compounds. 1.3 R <sub>f</sub> value, factors affecting R <sub>f</sub> value, detection, identification and estimation.	I	<ul><li>1.1 Thin Layer Chromatography-Principle, techniques, preparation of chromatoplates, application of sample on the chromatoplates.</li><li>1.2 Choice of adsorbents, selection of solvents, locating reagents, developing chamber, development types, location of compounds.</li></ul>			
II       Determination of Physical Constants         2.1. Melting point- Definition, determination by Thiele and Gallenkump melting point apparatus.       6         2.2. Boiling point – Definition and determination       6	п	2.1. Melting point- Definition, determination by Thiele and Gallenkump melting point apparatus.	6		
Thin Layer Chromatography- Practical-I63.1 Preparation of TLC plates63.2 Separation of pigments from a mixture of red ink, blue ink and black ink.	III	3.1 Preparation of TLC plates	6		
IVThin Layer Chromatography- Practical-IIIV4.1 Separation of mixture of amino acids4.2 Separation of mixture of phenols	IV	4.1 Separation of mixture of amino acids	6		
Determination of Physical Constants - Practical         5.1 Determination of melting point of the following solids         i) Benzoic acid         ii) Benzamide         V         iii) Naphthalene         5.2. Determination of boiling point of the following liquids:         i) Ethyl acetate         ii) Water         iii) Ethyl Methyl Ketone	V	<ul> <li>5.1 Determination of melting point of the following solids <ul> <li>i) Benzoic acid</li> <li>ii) Benzamide</li> <li>iii) Naphthalene</li> </ul> </li> <li>5.2. Determination of boiling point of the following liquids: <ul> <li>i) Ethyl acetate</li> <li>ii) Water</li> </ul> </li> </ul>	6		

#### **Text Book(s):**

1. B. K. Sharma, Instrumental Method of Chemical Analysis, First Paper Print Edition, Krishna Prakashan Media (P), Ltd, Meerut, Uttar Pradesh, 2023.

2. V. K. Srivastava and K. K. Srivastava, Introduction to Chromatography- Theory and Practice, 4<sup>th</sup> Edition, S. Chand and Company (Pvt) Ltd., New Delhi-110 002, 2020.

#### **Reference Book:**

1. H. Kaur, Instrumental Methods of Chemical Analysis, 12<sup>th</sup> Edition, Pragati Prakashan, Meerut, Uttar Pradesh, 2022.

### Web Resource(s):

- 1. https://archive.nptel.ac.in/content/storage2/courses/102103047/PDF/mod4.pdf
- 2. <u>https://www.youtube.com/watch?v=74vCYTIHo5Q</u>

Semester	Course Code	Course Category	Hours/ Week	Credits	Marks for Evaluation		
Semester					CIA	ESE	Total
III	24PCHVAC1	Value Added Course	30	-	-	100	100
Course Title CHROMATOGRAPHY AND PHYTOCHEMICAL ANALYSIS							

	SYLLABUS				
UNIT	CONTENTS	HOURS			
Ι	Paper chromatography-Definition-principle-Instrumentation-Techniques -Applications.	6			
II	Gas chromatography- Principle- Instrumentation- Techniques - Applications.	6			
III	Extraction Techniques- Practical           General methods of extraction –Percolation and Soxhlet extraction.				
IV	High performance liquid chromatography (HPLC) - PracticalPrinciple, Instrumentation, Techniques, Identification of chemical components ofNeem plant extract.	6			
v	Phytochemical Analysis - Practical Identification of alkaloids-Mayer's test, terpenoids-Salkowski test, flavonoids -lead acetate test, protein-Ninhydrin test, saponins-capillary test, tannins-Frothing test, glycosides-Keller-Killiani test and carbohydrates-Molisch's test.	6			

## **Text Book(s):**

1. Dhruba Charan Dash, Analytical Chemistry, PHI Learning Pvt Ltd, 2<sup>nd</sup>Edition, 2017.

 Gurdeep R Chatwal, Organic chemistry of Natural products, Volume I&II, Himalaya Publishing House, 5<sup>th</sup> Edition, 2019.

### **Reference Book(s):**

1. E. Hywel Evans and Mike E.S, Analytical Chemistry a Practical Approach, Oxford University Press, 1<sup>st</sup> Edition, 2019.

2. Jeffrey B. Harborne Phytochemical methods: A Guide to Modern Techniques of Plant Analysis, Indian Reprint, Springer, 5<sup>th</sup>Edition, 2013.

## Web Resources:

1. <u>https://www.youtube.com/watch?v=EvkZxx5if84</u>

## Course Coordinator(s):

Dr. A. Zahir Hussain, Dr. H. Mohamed Kasim Sheit, Dr. A. Samsath Begum